

UNIVERSITY OF NORTH BENGAL

B.Sc. Honours 2nd Semester Examination, 2023

CC3-GEOLOGY

ELEMENTS OF GEOCHEMISTRY

Time Allotted: 2 Hours

The figures in the margin indicate full marks.

- 1. Write short notes on any *five* of the following:
 - (a) Which mineral is most suitable for the geochronology?
 - (b) Name the most abundant element and mineral of earth crust.
 - (c) Why zircon is commonly used in mineral geochronology?
 - (d) Why does the e-process stops at mass number 56?
 - (e) Compare the properties of solid with Ionic and Covalent bonding.
 - (f) What type of meteorite is important in resolving the CHUR age-model?
 - (g) Describe the compatible and incompatible element.
 - (h) What are achondrites?

2. Answer any *three* questions from the following:

- (a) Write down the geochemical classification of elements with suitable examples.
- (b) A granitic rock sample has ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁷Sm/¹⁴⁴Nd of 0.51215 and 0.1342, respectively. The present chondritic ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁷Sm/¹⁴⁴Nd are 0.512638 and 0.1967, respectively. The decay constant of ¹⁴⁷Sm is 6.54×10^{-12} Ga⁻¹. Calculate the τ_{CHUR} , i.e., crustal residence time relative to a chondritic mantle.
- (c) Classify magma with respect to SiO_2 percentage. Which type of magma is least viscous?
- (d) Write down a short note on Chondrite and stony-iron meteorite.
- (e) Differentiate between the Enriched Mantle and Depleted Mantle.

3. Answer any *two* questions from the following:

- (a) State Pauling's principles for ionic substation. What is the role of Gibbs free 4+6 energy (G) in determining the stability of mineralogical assemblages? Express G from the laws of thermodynamics.
- (b) What is fractionation factor in stable isotope notation? Why Isotopic fractionation is more pronounced in lower temperature? Write down a short note on Isochron method.
- (c) With the help of Harkar Diagram explain the general trends of Major and Minor oxides in an evolving Magma. Give suitable diagrams to illustrate your answer.
- (d) Describe the differences in the analytical methods used in EPMA and XRF and the specific purposes for these methodologies.

Full Marks: 40

 $1 \times 5 = 5$

 $5 \times 3 = 15$

 $10 \times 2 = 20$

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